



Article Discontinuous Geochemical Monitoring of the Galleria Italia Circumneutral Waters (Former Hg-Mining Area of Abbadia San Salvatore, Tuscany, Central Italy) Feeding the Fosso Della Chiusa Creek

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Abstract: The Galleria Italia waters drain the complex tunnel system of the former Hg-mining area of Abbadia San Salvatore (Tuscany, central Italia) and feed the 2.5 km-long Fosso della Chiusa creek. The mining exploitation was active for more than one century and more than 100,000 tons of liquid mercury were produced by roasting processes of cinnabar (HgS). In this work, a discontinuous geochemical monitoring of the Galleria Italia circumneutral waters was carried out from February 2009 to October 2020, during which the main physicochemical parameters, main and minor dissolved species and trace elements (including Hg) were determined. In the observation period, significant variations in the water chemistry were recorded, particularly when flooding waves, due to intense precipitations, occurred, with the two main events being recorded in February 2009 and January 2010. The chemical composition of the Galleria Italia waters was Ca(Mg)-SO4 and related to congruent dissolution of gypsum/anhydrite at which a contribution from carbonatic and silicatic minerals and partial solubilization of CO₂ and and H₂S oxidation is to be added. Regarding the trace elements, Al, Mn and Fe were up to 1500, 768 and 39520 μ g L⁻¹, with these elements also showing high contents in the sediment precipitating by the Galleria Italia waters. In most cases, dissolved mercury was below the instrumental detection limit (<0.1 μ g L⁻¹), although occasionally it reached >1 μ g L⁻¹. Considering a mean flow rate of 40 L s^{-1} of the discharged water, the amount of dissolved mercury released from Galleria Italia was computed, although most mercury was occurring in the sediment (1.2 mg kg^{-1}) . A more realistic computation of mercury released from Galleria Italia should involve a sampling network along the Fosso della Chiusa before entering the riverine system of the Tiber basin, into which dissolved and suspended mercury are to be determined along with that occurring in the sediments.

Keywords: water geochemistry; mercury; heavy metals; Abbadia San Salvatore; central Italy; abandoned mining areas; circumneutral waters

1. Introduction

Mine drainage is mostly considered to be acidic (pH < 5) although circumneutral (pH between 6 and 8) to basic (pH > 8) waters are also found. The redox and pH conditions of



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Copyright: © 2021 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (https:// creativecommons.org/licenses/by/ 4.0/). these waters are expected to control the concentrations of toxic metals, mostly heavy metals and metalloids. Acidic Mining Drainage (AMD) has received a lot of attention, likely due to the sometimes devastating effects on the environment resulting from the oxidation of pyrite and other poly-metallic sulfides (e.g., [1–4]). Pyrite tends to be altered at surface conditions, as follows (e.g., [5])

$$FeS_{2(s)} + \frac{15}{4O_{2(aq)}} + \frac{7}{2H_2O} = Fe(OH)_{3(s)} + \frac{2SO_4^{2}}{(aq)} + 4H^{+}_{(aq)}$$

Conversely, circumneutral mine waters (e.g., [6] and references therein) have been less investigated. They can be related to interaction processes between meteoric waters and rocks containing low contents of metal- and metalloid-sulfides. The neutralizing agents of the acidic waters resulting from sulfide oxidation are carbonate- (e.g., CaCO₃ and MgCO₃) and silicate-bearing (e.g., feldspars) rocks (e.g., [7])

$$CaCO_{3} + H^{+} = HCO_{3}^{-}{}_{(aq)} + Ca^{2+}{}_{(aq)}$$
$$CaAl_{2}Si_{2}O_{8(s)} + 2H^{+}{}_{(aq)} + H_{2}O = Ca^{2+}{}_{(aq)} + Al_{2}Si_{2}O_{5}(OH)_{4(s)}$$

Thus, mixing between acidic and circumneutral waters or water–rock interactions with low sulfide-bearing silicatic rocks produce a slightly acidic or neutral pH that favors the precipitation of metals as oxy-hydroxides and gypsum. Consequently, circumneutral waters contain lower concentrations of toxic elements than those recorded in acidic waters. Nevertheless, trace elements can be dissolved at slightly higher or near to acceptable concentrations whereas those of sulfate may remain high (e.g., [4,8,9]), thus requiring decontamination treatment technologies (e.g., [10,11]). Among the trace elements that can be found dissolved in circumneutral waters, arsenic and, to a lesser amount, antimony showed concentrations up to mmol L^{-1} ([12]). According to [13], mining waters with pH approaching neutrality can indeed be of environmental concern, since the mobilization of chalcophile elements can occur.

To the best of our knowledge, few studies relate to the concentration of mercury in circumneutral waters (e.g., [8,10,11]), and intending, at least partially, to fill this gap, in this work, we have analyzed the waters discharged from Galleria Italia that represent the draining waters of the tunnels drilled in the former Hg-mining area of Abbadia San Salvatore (Tuscany, central Italy), from which cinnabar was exploited to produce liquid mercury. Since these meteoric-fed waters showed variations in terms of flow rate, from 2009 to 2020, periodic discontinuous samplings were carried out to evidence possible chemical changes by investigating the main physicochemical parameters and geochemistry and selected trace elements, including mercury. The Galleria Italia waters are the main supply of the Fosso della Chiusa, a small creek that enters the river network of the Tiber basin. Consequently, evaluating the presence of mercury and other toxic elements is of environmental importance.

2. Geological Outlines and the Study Site

The Mt. Amiata mining district was one of the most important areas worldwide for the production of mercury, whose activity lasted for more than one century (1847–1976). The Hg(cinnabar)-ore deposits were found in both sedimentary formations and volcanic rocks, the latter being related to the 200–300 ka old Mt. Amiata volcano. The origin of the ore deposits is still matter of debate, although [14–17] argued that a complex mobilization process of mercury (and antimony) affecting the Paleozoic phyllites occurred between the Late Oligocene and Early Miocene. Then, a second mobilization (Pliocene to Pleistocene) was invoked for the formation of the Mt. Amiata epithermal deposits. According to [18], the mobilization/deposition process is still ongoing. At least 15 sites in the Mt. Amiata mining district were operating to exploit cinnabar (e.g., [19]), but that of Abbadia San Salvatore (Figure 1a) was by far the most important. It was estimated that more than 100,000 tons of liquid mercury were produced (e.g., [20]). The environmental impact due to the production of mercury has left most of the geological and biological compartments, as well as the mining structures, polluted (e.g., [21-43]). Surface and ground waters were evidenced to be particularly affected by the presence of high mercury concentrations (e.g., [36] and references therein). The waters from Galleria Italia $(42^{\circ}53'4.9'' \text{ N}, 11^{\circ}40'32'' \text{ E};$ emerging at 786.50 m a.s.l.) are discharged slightly below the urban center of Abbadia San Salvatore. They play a key role in the management of the mining waters since they are convoyed to Galleria Italia (Figure 1a,b). In the past, there was another water discharge system at the Galleria Ribasso-200 m (42°53'17" N, 11°41'53" E; altitude 511.50 m a.s.l.), which was not functioning when the water monitoring at Galleria Italia started in 2009. It was suggested that the collapse of the Galleria Ribasso-200 m is preventing any outflow at the surface [44]. To date, a schematic geological section with the hydrogeological circuit of the mining tunnels in the former mining area of Abbadia San Salvatore is reported in Figure 2. The mean flow rates, measured by the personnel of the local municipality, are also reported. It is notable that, from 2009 to 2020, the flow rates of the Galleria Italia waters showed a mean value of $40 \text{ L} \text{ s}^{-1}$, although they ranged from 20–30 up to 110 L s^{-1} [44]. A few meters after the emergence, half of the flow rate is diverted to feed a small hydroelectric plant, named "La Turbina".



Figure 1. Location of Abbadia San Salvatore: (**a**) Galleria Italia and (**b**) a detail of the water discharge where the reddishwhitish color can be observed.

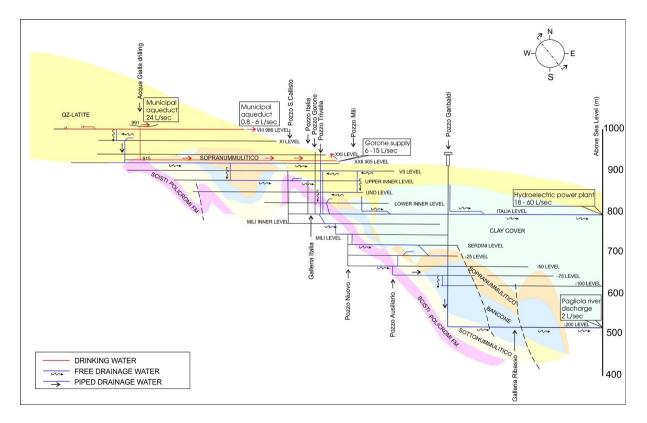


Figure 2. Schematic geological cross-section of the mining tunnels in the former mining site of Abbadia San Salvatore. For the sake of clarity, the local geological terms, such as scisti policromi (shales), sotto- and sopra-nummilitico (alternated calcarenites with grayish-greenish clays and grayish-pinkish calcarenites with grayish clays) and bancone (massive calcarenites) have been maintained (modified after [44]).

No direct access to the mining tunnel was possible, since it was cemented after the closure of the mining activity and only few nozzles were left, to allow the water to flow through [45]. According to the miners, thermal waters (ca. 35–37 °C) were observed and they were mixed with the main drainage system before emergence (M. Niccolini, pers. comm.). The Galleria Italia waters, which are reddish-whitish in color, feed the Fosso della Chiusa creek, which, in turn, enter the Pagliola River after about 2.5 km. Eventually, these surface waters reach the Paglia River, and then the Tiber River [46]. This implies that variations in the physicochemical parameters and the water chemistry can be regarded as diagnostic of the geochemical processes occurring inside the underground mining structures and a good proxy for evaluating the effects of potential contaminants entering the riverine network. During the monitoring activity, three important events occurred: (i) a flow wave on the 12th of February, 2009 (a similar phenomenon was reported to occur in 2006); (ii) an abrupt change in the water color on the 17th of February, 2009 when the waters turned from a reddish color, typical of the presence of suspended Fe-oxy-hydroxides, to milky reddish-whitish, indicating the presence of suspended Al-compounds, as also suggested by the chemical composition of the precipitating material; (iii) a flow wave on the 10th of January, 2010, though less intense than that of February 2009. Further details on these events, mostly related to extraordinary precipitations that provoked a flow rate increase >110 L s⁻¹, are reported in [44].

3. Sampling and Analytical Methods

From January 2009 to October 2020, 42 water samplings were carried out at the emergence of Galleria Italia. The water turbulently outflows and enters after 70 cm in a small (80 cm wide) channel. The water (about 20 cm thick) rapidly flows away in a turbulent regime and feeds the Fosso della Chiusa creek. Temperature, pH and electrical

conductivity were measured in situ with portable instrumentations (Crison MM40+ and HI98194, respectively; HACH, Barcelona, Spain). During each sampling, four aliquots were collected: (i) 0.45 mm filtered sample in 125 mL PE bottles equipped with counter-cap for anions and NH_4^+ ; (ii) 0.45 µm filtered and acidified (1% suprapur HCl) sample in PE 50 mL bottles for Na, K, Ca and Mg; (iii) 0.45 µm filtered and acidified (1% suprapur HNO₃) sample in PE 50 mL bottles for trace elements (Al, B, Fe, Mn, Co, Cr, Ni, As, Sb, Zn and Sr); (iv) 0.45 µm filtered and acidified (1% suprapur HCl) in 75 mL dark glass bottles for mercury. The bottles were cleaned prior to sampling several times with MilliQ water and dilute HCl. Blank analyses produced Hg concentrations $< 0.1 \mu g L^{-1}$. The sediment, <10 cm thick and reddish in color, was collected at the water emergence with a plastic scoop and transferred to a plastic bag and analyzed for some major (Na, K, Ca, Mg, Al, Fe) and trace elements (Mn, Cr, Ni, Zn, As, Sb and Hg). Bicarbonates were measured by acidimetric titration with 0.01 N HCl and methyl-orange as indicators. The main anions and cations were determined by ion-chromatography using a 761 Compact IC and an 861 Advanced Compact IC (Metrohm, Herisau, Switzerland), respectively. Ammonium was measured by molecular spectrophotometry (DR2010 HACH, Barcelona, Spain) by the Nessler method. The analytical error for these dissolved species was < 3%. Trace elements in water were analyzed by ICP-MS (Model 7500CE Agilent, Santa Clara, USA; method: EPA 6020B 2014) at the CSA Research Group of Rimini (Italy), accredited by group ACCREDIA. Appropriate internal standards, e.g., ⁶Li, ⁴⁵Sc, ⁸⁹Y and ¹¹⁵In, were used to set up the ICP-MS instruments. Specific details on the ICP-MS set up are reported in [47,48]. The QA/QC protocol for ICP-MS is reported in Table S1 (Supplementary Materials). In the laboratory, the sediment, consisting of an impalpable powder, was dried at 40 °C. Then, 0.5 g were digested in aqua regia and the trace elements were analyzed by ICP-AES (720ES Agilent, Santa Clara, USA; method EPA 200.7). Dissolved and total mercury was determined by DMA-80 (Direct Mercury Analysis; Milestone, Sorisole (BG), Italy), in full compliance with EPA method 7473. DMA allows to determine Hg by thermal decomposition, mercury amalgamation and atomic absorption spectrometry with pure (>> 99.999%) O_2 as a carrier (e.g., [49] at the CSA Research Group of Rimini (Italy). The analytical error was <10%.

4. Results

4.1. Water Geochemistry

Temperature (in $^{\circ}$ C), pH and the concentration (in mg L⁻¹) of the main and minor dissolved species determined in the last 12 years for the Galleria Italia waters are reported in Table 1, along with the sampling date, Total Dissolved Solids (TDS). The electroneutrality parameter calculated according to [50] was always < 5%. Not all the geochemical parameters were determined for the waters collected in September 2017, January 2018, November 2018, May 2019 and January 2020 and they were only partially analyzed (Table 1). Each sample was identified with a progressive number and these numbers were used in the geochemical diagrams. The minimum, maximum and mean values of temperature, pH and TDS (in mg L^{-1}) were 7.6, 18 and 14.8 °C, 5.45, 6.99 and 6.16 and 757, 1567 and 962 mg L^{-1} , respectively. In most cases, the lower the pH the higher the TDS values, which is likely related to a higher leaching capacity and a higher concentration of dissolved CO₂ [44]. Moreover, the highest TDS value was recorded a few days after the February 2009 flooding wave. According to the cation and anion triangular diagrams (by recalculating the concentrations in meq L^{-1} to 100%) reported in Figure 3, and the pH values, the Galleria Italia waters can be classified as circumneutral and Ca(Mg)-SO₄ in composition. The Ca and SO₄ concentrations varied from 180 and 342 mg L^{-1} and 403 and 996 mg L^{-1} , respectively, while those of Mg were between 16.5 and 48 mg L^{-1} . HCO₃, the second most abundant anion, had concentrations between 21 and 213 mg L^{-1} . Notably, the lowest contents of bicarbonates were recorded after the flood events and they started to recover after December 2009. Na, K and Cl were always below 20, 37 and 26 mg L^{-1} , respectively. Finally, NH₄, F and NO₃ never exceeded 9.3, 2.02 and 2.8 mg L^{-1} , respectively.

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	Sampling	Т	pН	TDS	Ca ²⁺	Mg ²⁺	Na ⁺	K ⁺	$\mathrm{NH_4^+}$	HCO ₃ -	SO4 ²⁻	Cl-	\mathbf{F}^-	NO_3^-
ID	Date	°C		mg L ⁻¹	${f mg} \ {L^{-1}}$									
1	30/01/2009	12.6	5.54	1017	212	31	14	11	4.8	138	595	9.9	1.2	0.5
2	12/02/2009		5.90	1567	342	48	20	16	n.d.	136	996	8.7		0.5
3	18/02/2009	12.0	6.57	1178	222	37	14	37	8.2	136	697	26.0	0.5	0.4
4	23/03/2009			992	227	35	15	12	0.2	21	674	8.3		
5	26/03/2009	16.0	5.45	1034	211	33	13	12	5.7	21	732	4.5	1.0	0.4
6	24/07/2009	17.5	5.96	917	196	27	14	12	3.6	38	618	6.2	0.9	0.9
7	30/10/2009	17.0	5.95	914	195	27	13	11	1.0	85	576	5.3	1.0	0.0
8	22/12/2009	15.5	5.97	881	192	25	13	11	3.5	111	517	6.5	0.6	1.4
9	11/01/2010		5.62	1061	235	27	12	10	0.6	100	670	5.7	0.4	0.7
10	14/01/2010	12.0	5.83	1217	289	35	10	11	9.3	155	700	5.3	0.8	1.7
11	06/04/2010	15.5	5.48	1019	213	35	13	11	1.5	44	694	6.1	0.6	0.4
12	21/07/2010	15.5	5.51	1002	231	30	15	10	7.1	35	663	7.5	0.8	2.8
13	21/11/2010	16.5	6.09	952	201	24	13	9	2.4	88	608	5.0	0.6	1.0
14	14/01/2011	15.0	6.21	1038	234	28	13	10	3.4	134	609	6.1	0.7	0.1
15	21/04/2011	16.0	5.88	1037	227	29	12	10	0.5	107	630	19.0	1.4	0.8
16	29/06/2011	16.5	5.90	946	200	25	12	11	0.3	120	560	16.0	1.0	0.4
17	14/10/2011	15.7	6.19	949	215	25	13	12	0.1	127	541	15.0	0.8	0.2
18	17/11/2011	14.5	6.54	938	222	26	15	12	0.4	110	533	17.0	1.8	1.1
19	28/12/2011	15.0	6.76	948	223	26	13	11	0.5	139	526	8.5	0.9	0.2
20	08/03/2012	15.0	6.58	938	213	25	13	10	0.1	213	447	15.0	1.0	0.6
21	28/04/2012	11.5	6.14	953	216	25	14	12	0.1	175	493	16.0	1.1	1.1
22	17/07/2012	15.5	6.26	897	197	22	12	10	0.1	166	473	16.0	0.8	0.1
23	12/10/2012	15.5	6.20	915	200	23	13	9	0.1	156	496	16.0	0.8	0.8
24	03/01/2013	15.0	6.30	959	211	29	13	10	0.1	120	557	17.0	1.0	0.9
25	21/05/2013	15.5	5.80	788	191	17	11	10	0.0	74	475	10.0	0.7	0.1
26	12/02/2014	14.2	6.06	1036	238	33	12	12	0.2	151	574	16.0	0.7	0.1
27	10/09/2014	15.0	6.18	816	194	18	11	10	0.0	94	477	11.8	0.8	0.1
28	16/04/2015	14.5	6.20	911	202	23	9	8	0.1	115	542	10.7	0.6	0.2
29	24/10/2015	15.0	6.19	852	206	21	11	9	1.2	111	478	13.7	0.4	0.2
30	03/05/2016	15.0	5.98	917	220	24	12	10	0.4	125	512	13.0	0.5	0.4
31	28/10/2016	12.6	5.83	849	198	22	15	10	0.2	120	475	8.0	0.5	0.1
32	04/04/2017	14.9	6.37	871	192	23	14	11	0.2	192	428	10.0	1.1	0.2
33	01/09/2017	15.0	6.20											
34	10/01/2018	13.7	6.99											
35	01/06/2018	17.5	6.47	854	180	26	12	13	0.0	78	530	13.0	2.0	0.1
36	01/11/2018	7.6	6.54	001	100			10	0.0		000	10.0		0.1
37	01/03/2019	14.5	6.58	917	192	27	13	11	1.3	149	511	12.4	0.8	0.1
38	16/05/2019	14.8	6.60	/1/	1/2		10		1.0	11/	011	12.1	0.0	0.1
39	01/10/2019	17.0	6.60	824	180	21	11	9	0.1	92	498	12.0	0.1	0.1
40	07/01/2020	13.0	6.41	U= 1	100		**	/			270		0.1	
41	01/05/2020	15.4	6.52	941	206	23	12	8	3.5	104	558	25.1	1.2	0.1
42	01/10/2020	16.0	6.28	757	181	23	8	5	0.5	114	404	22.3	1.2	0.1
	51, 10, 2020	10.0	0.20		101	-1	0	0	0.0		101		1.4	

Table 1. Temperature, pH and concentration of the Total Dissolved Solids (TDS), main and minor dissolved species in the Galleria Italia waters. The Identification Numbers (IDs) are those used in the diagrams to refer to the sampling date.

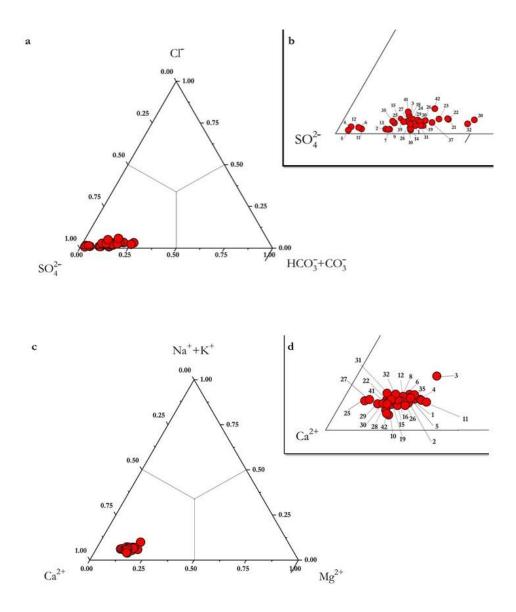


Figure 3. Triangular diagrams of Cl-HCO₃ + CO₃-SO₄ (**up**, **diagrams a and b**) and Na + K-Mg-Ca (**down**, **diagrams c and d**). For each diagram, an inset is included (respectively b and d) to better observe the distribution of the analyzed samples from Galleria Italia. The Identification Numbers (IDs) refer to refer to the different surveys (see Table 1).

The selected trace element concentrations (in μ g L⁻¹), ordered according to the increasing atomic number, are listed in Table 2. Despite the circumneutral pH values, high concentrations of Fe, Mn, and Al (up to 39,520; 768 and 1500 μ g L⁻¹) were measured. When measured, the Sr contents were always above 176 μ g L⁻¹. Boron, zinc, cobalt and nickel varied between 20.5 and 96.5, 33.6 and 244, 4.3 and 26.3, and 28 and 86 μ g L⁻¹, respectively. In most cases, Cr and Sb were below the instrumental detection limit (0.1 μ g L⁻¹), while those of As were comprised between 5.9 and 23.4 μ g L⁻¹. Finally, Hg showed concentrations up to 25.6 μ g L⁻¹, although they were < 0.1 μ g L⁻¹ in 15 samples out 38 (Table 2) and only eight samples had concentrations > 1 μ g L⁻¹.

ID -	Sampling	В	Al	Cr	Mn	Fe	Со	Ni	Zn	As	Sr	Sb	Hg
10	Date	$\mu g/L^{-1}$											
1	30/01/2009	82	712	< 0.1	768	26,079	23	70	220	16	792	< 0.1	7.1
2	18/02/2009	63	166	< 0.1	740	34,027	24	76	195	16	722	< 0.1	0.9
5	26/03/2009	75	1498	< 0.1	689	39,521	25	86	244	23	629	< 0.1	0.1
6	24/07/2009	59	906	< 0.1	530	23,003	17	65	171	11	522	< 0.1	< 0.1
7	30/10/2009	76	651	< 0.1	682	22,179	22	63	161	10	623	< 0.1	< 0.1
8	22/12/2009	49	346	< 0.1	601	20,926	14	44	121	9	580	0.4	0.5
10	14/01/2010	42	252	< 0.1	616	21,930	19	59	109	6	837	0.1	< 0.1
11	06/04/2010	74	1110	< 0.1	687	32,260	26	86	217	12	747	0.1	0.2
12	21/07/2010	60	1403	1.4	732	28,572	25	74	209	14	630	< 0.1	0.1
13	21/11/2010	59	331	< 0.5	528	19,622	16	51	126	10	461	< 0.1	1.4
14	14/01/2011	50	387	0.2	468	15,656	14	52	100	6	468	< 0.1	0.2
15	21/04/2011	61	865	< 0.1	567	20,063	18	73	183	13	519	< 0.1	< 0.1
16	29/06/2011	68	731	< 0.1	536	16,476	15	51	140	14	541	< 0.1	< 0.1
17	14/10/2011	65	421	0.3	490	14,960	12	43	115	14	487	< 0.1	< 0.1
18	17/11/2011	33	93	0.4	461	17,920	4	39	34	9	176	< 0.1	0.4
19	28/12/2011	70	110	0.1	475	15,920	11	37	83	7	476	< 0.1	13.2
20	08/03/2012	82	179	0.1	502	17,380	10	37	89	7	496	< 0.1	3.5
21	28/04/2012	97	320	0.2	436	13,410	9	33	84	9	459	< 0.1	0.2
22	17/07/2012	55	283	0.2	452	11,253	9	29	85	6	453	< 0.1	25.6
23	12/10/2012	49	275	0.1	451	12,384	8	28	90	6	455	< 0.1	9.0
24	03/01/2013	43	134	< 0.1	454	13,240	11	44	134	8	524	< 0.1	7.0
25	21/05/2013	63	926	0.2	486	20,060	16	63	177	13	560	< 0.1	< 0.1
26	12/02/2014	52	660	0.2	470	18,130	12	47	135	13		< 0.1	5.7
27	10/09/2014	45	409	0.1	352	10,060	14	40	116	11		< 0.1	< 0.1
28	16/04/2015	21	149	0.1	364	13,560	12	43	100	11		0.5	< 0.1
29	24/10/2015	44	183	0.1	352	12,050	11	30	100	7		< 0.1	< 0.1
30	03/05/2016	43	346	0.1	427	15,000	11	43	124	9		< 0.1	< 0.1
31	28/10/2016	80	413	< 0.1	424	14,861	9	38	106	8		< 0.1	< 0.1
32	04/04/2017	33	323	0.1	345	20,290	7	30	98	10		< 0.1	< 0.1
34	10/01/2018									8		0.6	< 0.1
35	01/06/2018	34	346	0.2	411	32,100	11	44	158	12			0.2
36	01/11/2018					,				9		< 0.1	0.4
37	01/03/2019	42	270	0.2	404	18,270	8	37	112	11		0.3	< 0.1
38	16/05/2019									9		< 0.1	< 0.1
39	01/10/2019									11		< 0.1	0.4
40	07/01/2020	40	396	0.2	377	5876	11	39	119				0.2
41	01/05/2020		•							11		< 0.1	2.2
42	01/10/2020									8		<0.1	1.0
										č			1.0

Table 2. Trace element concentrations of the Galleria Italia waters. The progressive numbers are those used in the diagrams to identify the sampling date. The sampling 3, 4, 9 and 33 are not listed because no data were available for trace elements. The Identification Number (IDs) are those used in the diagrams to refer to the sampling date.

4.2. Sediment Geochemistry

The concentration of the main (Na, Mg, Al, K, Ca and Fe) and trace (Cr, Mn, Ni, Zn, As, Sb and Hg) elements in the sediment precipitating from the Galleria Italia waters is reported in Table 3 and they refer to the mean values of three sampling campaigns carried out between 2009 and 2010. Iron and Al showed by far the highest contents, since they were of 37.20 and 4.42 wt.% whereas Na, K, Ca and Mg were below 0.7 wt.%. Mn, As and Zn were characterized by contents of 257, 192 and 191 mg kg⁻¹, respectively, whereas Cr, Ni and Sb showed concentrations of one order of magnitude less than those of the other analyzed trace elements, with the exception of that of mercury, whose content was of 1.2 mg kg⁻¹.

Element	Unit of Measurement	Concentration				
Na	%	0.02				
Mg	%	0.07				
AÌ	%	4.42				
K	%	0.07				
Ca	%	0.7				
Fe	%	37.2				
Cr	$ m mgkg^{-1}$	12				
Mn	$mg kg^{-1}$	257				
Ni	$mg kg^{-1}$	16				
Zn	$mg kg^{-1}$	191				
As	$mg kg^{-1}$	192				
Sb	$mg kg^{-1}$	13				
Hg	$mg kg^{-1}$	1.2				

Table 3. Major (wt.%) and trace element (mg kg $^{-1}$) in the sediment precipitating from the Galleria Italia waters.

5. Discussion

5.1. Origin of Solutes

The Galleria Italia waters can be classified as belonging to the circumneutral waters, with the pH values clustering around 6.2 and characterized by a Ca(Mg)-SO₄ geochemical facies (Figure 3), as also evidenced by the (Ca + Mg) + SO₄ vs. TDS (in mg L⁻¹) diagram (Figure 4). In most cases, these ions represent > 80% of the TDS, whereas the other main ions (HCO₃, Cl, Na and K) are subordinate. These chemical features are similar to those recorded for the surface and ground waters occurring inside the former mining area of Abbadia San Salvatore and whose origin was referred to water–rock interaction processes where relatively soluble minerals (e.g., sulfates and, at minor extent, carbonates) are involved along with a silicate component [36]. When SO₄ (in meq L⁻¹) is subtracted to the sum of Ca + Mg (in meq L⁻¹), most water samples are stoichiometrically associated with HCO₃, suggesting that carbonate dissolution also contributes to the geochemical features of the Galleria Italia waters. The Na/Cl ratio (in meq L⁻¹) is always higher than the stoichiometry and seawater ratios, thus indicating the presence of a Na-silicate contribution.

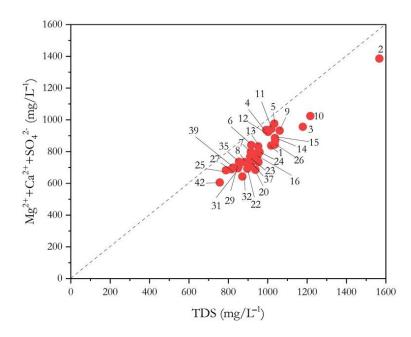


Figure 4. Mg + Ca + SO₄ (in mg/L⁻¹) vs. TDS (in mg L⁻¹) binary diagram. The IDs refer to the different surveys (see Table 1). The dotted line represents the (Mg + Ca + SO₄)/TDS = 1.

According to [44], sulfate can also be related to both oxidation processes of polymetallicsulfides and H₂S, as supported by the δ^{34} S-SO₄ values and alteration processes of calcine and slags operated by meteoric waters feeding the Galleria Italia waters [36]. As observed by [51,52], the Mt. Amiata waters can be distinguished into four groups: (i) low TDS (<250 mg L⁻¹) waters with (Na + K)/(Ca + Mg) and (HCO₃)/(Cl + SO₄) ratios close to 1 and circulating in the Mt. Amiata volcanics; (ii) Ca-HCO₃ waters with TDS of about 400 mg L⁻¹ and associated with Mesozoic carbonate and Oligo-Miocene turbidites; (iii) waters with TDS > 2000 mg L⁻¹ and temperatures up to 50 °C [53] and belonging to a Ca(Mg)-SO₄(HCO₃) composition, since they circulate in evaporitic–carbonatic formation; iv) SO₄-rich acidic waters. In addition, Na-Cl were found to characterize the geothermal fluids exploited in the eastern-southern part of Mt. Amiata [51].

The Galleria Italia waters, characterized by low contents of Na and Cl (Table 1), can be regarded as compositionally intermediate between the previously reported first three types of waters. This composition could suggest that a thermal component occurring inside the mining tunnels, as reported by some miners, is likely.

As previously mentioned, relatively high concentrations of Fe, Mn and Al were measured in both the Galleria Italia waters and sediments (Tables 1 and 2). Normally, these elements tend to be immobile during chemical alteration processes. However, such high contents can be indicative of relatively acidic and/or reductive conditions inside the Galleria Italia to favor the formation of hydroxylate complexes or, with the sulfate ion, allowing the presence of these elements in solution even when dilution processes with shallower waters occur. It can be suggested that the relatively slow kinetic processes do not prevent their precipitation before the Galleria Italia waters emerge at the surface. This is supported by their high contents in both the water and the sediment. The As behavior is peculiar since it tends to be strongly enriched in the solid phase (Table 3), although in solution the concentration is often higher than 10 μ g L⁻¹, which is the maximum permissible concentration for drinkable waters.

It is notable that the access inside the Galleria Italia is presently not viable and, consequently, the geochemical processes governing the composition of the circumneutral drainage waters of the former mining area of Abbadia San Salvatore can only be hypothesized. Fe, Al, Mn and As can, however, be related to a common origin (e.g., [54,55]), as also shown in Figure 5a–c.

As far as the relatively high contents of Ni and Co in the Galleria Italia waters, which are characterized by a positive correlation (Figure 5d), are concerned, they are likely due to dissolution processes of Ni-sulfides such as millerite and vaesite (e.g., [19] and references therein), while Zn concentrations may be related to the presence of sphalerite. Finally, Sr tends to replace Ca, and thus its origin can be ascribed to gypsum/anhydrite dissolution, while the contents of boron are a further clue to the presence of thermal waters, likely discharging into the drainage waters inside the Galleria Italia [56].

According to the chemical data reported in Tables 1 and 2, from 2009 to 2020 significant variations are recorded. Since there are no data about the flow rates during the monitored period, to verify the temporal variations and highlight possible significant changes due to either climatic events, such as those occurring in February 2009 and January 2010, or modifications in terms of relationships between the deep and shallow component, the analytical ratios (in meq L⁻¹) were then considered. The latter are not affected by flow rate variations, although the mean flow rate was of about 40 L sec⁻¹ (F. Piccinelli, pers. comm.). The temporal variability of the Ca/SO₄ and HCO₃/SO₄ ratios is reported in Figure 6, where Ca is mostly related to dissolution processes of sulfate and carbonate minerals and SO₄ may record congruent dissolution of gypsum/anhydrite and sulfide oxidation. HCO₃ can be regarded as derived by both the dissolution of carbonate phases and CO₂. In the diagrams of Figure 6, the February 2009 and January 2010 events are also evidenced. The considered ratios move in a synchronous way, i.e., the higher the Ca/SO₄ ratio the higher that of HCO₃/SO₄. The flooding events do not seem to affect the considered ratios since, during the February 2009 event, the Ca/SO₄ and HCO₃/SO₄ ratios are much lower than

those recorded after December 2011 and, with a few exceptions, April 2017. The Ca/SO₄ and HCO_3/SO_4 ratios show a tendency to increase with time after December 2009 although no significant pH variations were observed (Table 1). No indicative relationships were observed in terms of trace elements (including mercury). This decoupling may suggest that inside the Galleria Italia, an interplay between deep and shallow waters (both of meteoric origin, e.g., [44]) could govern the observed chemical variations. Nevertheless, more information about the geochemical processes can only be gathered by following the water path inside the Galleria Italia.

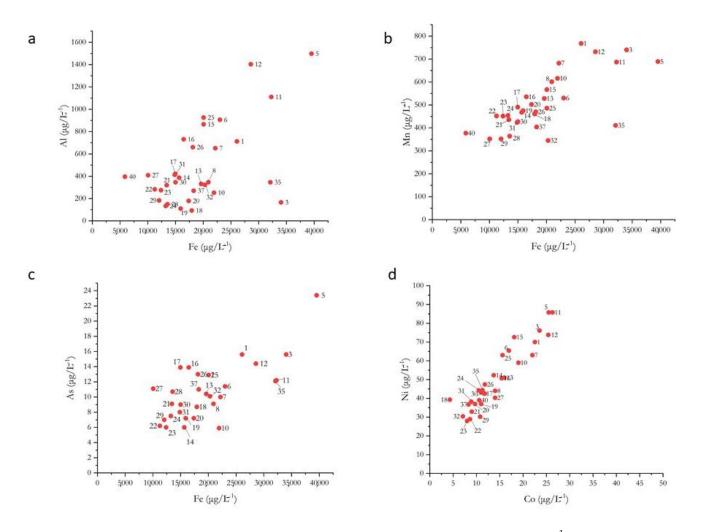


Figure 5. Binary diagram of Fe vs. Al (**a**), Mn (**b**) and As (**c**) and Ni vs. Co. All values are in mg L⁻¹. The numbers refer to the different sampling surveys (see Table 2). Statistical data for: (**a**) $\mathbf{r} = 0.5791$, p = 0.0005; (**b**) $\mathbf{r} = 0.7328$, p = 0.0001; (**c**) $\mathbf{r} = 0.7076$, p = 0.00001; (**d**) $\mathbf{r} = 0.9326$, p = 0.0001.

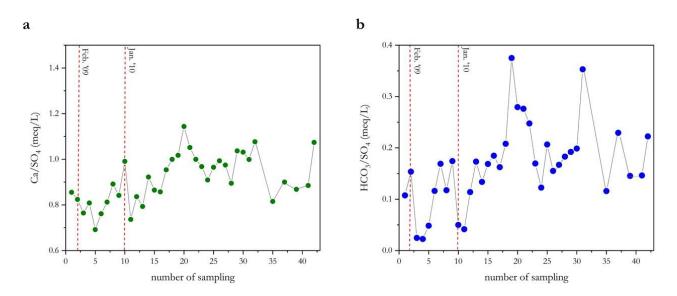


Figure 6. Temporal variations of the Ca/SO₄ (**a**) and HCO₃/SO₄ (**b**) ratios (in meq L^{-1}). The IDs refer to the different surveys (see Table 1).

5.2. Mercury

As mentioned, the concentrations of mercury in solution (Table 2) are relatively variable although, in most cases, the contents were below the instrumental detection limit. Only occasionally were they higher than the maximum permissible concentration $(1 \ \mu g \ L^{-1})$ for drinkable water. However, it is a matter of fact that most mercury is stored in the sediments (Table 3). Nevertheless, it was possible to compute the amount of dissolved mercury discharged from Galleria Italia by considering a reference value of $1 \ \mu g \ L^{-1}$ and a flow rate of 40 L s⁻¹. According to this computation the amount of dissolved mercury discharged from Galleria Italia is of about 1.26 kg y⁻¹. It is clear that this Hg load may be decreasing along the 2.5 km long Fosso della Chiusa creek, since mercury may be increasing when higher concentrations were recorded although, according to our data, they tend to decrease in a relatively short time. However, this calls for a detailed investigation of the dissolved and suspended mercury to be correlated with that occurring in the sediments [57].

If the computed dissolved mercury released from Galleria Italia is compared to those calculated by [33] for the surrounding rivers of the Mt. Amiata mining district, our dissolved load is of orders of magnitude lower. This observation fits with the circumneutral waters of Galleria Italia when the (Hg + As + Sb) vs. pH binary diagram ([58] and references therein) is considered (Figure 7), since near-neutral waters tend to be less enriched in these chalcophile elements with respect to the typical waters from acidic mine drainage.

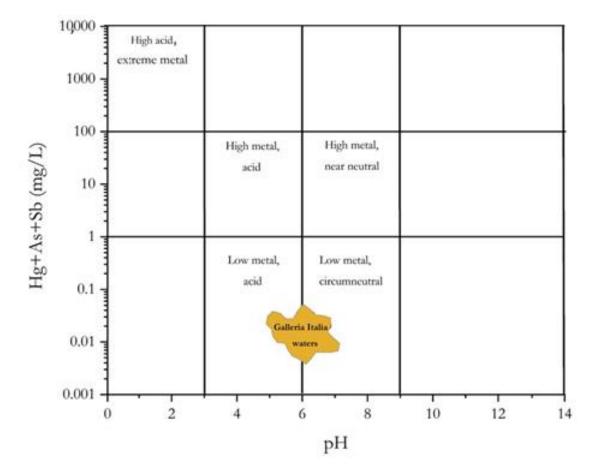


Figure 7. Hg + As + Sb (in mg L^{-1} in log scale) vs. pH (after [58]) for the Galleria Italia waters.

6. Conclusions

The waters from Galleria Italia represent the main drainage system of the mining tunnels of the former Hg-mining area of Abbadia San Salvatore. Consequently, they are the only way to possibly understand the geochemical processes occurring inside the exploited HgS-ore deposits. The infiltrating meteoric waters are indeed interacting with sulfide and carbonate-sulfate and silicate minerals, with the latter phases likely responsible for the circumneutral waters (Figure 7) recorded at the emergence of Galleria Italia. Nevertheless, the high concentrations of Fe, Al and Mn suggest that acidic waters may be occurring in the inner portions of Galleria Italia and buffered by alkaline hydrolysis processes. From 2009 to 2020, a relatively high chemical variability was recorded, suggesting that important changes were occurring due to climatic events and/or different degrees of mixing between a deep (thermal waters) and shallow (meteoric water interacting with the ore deposit) component. The concentrations of mercury were also variable although, in most cases, its content was $< 1 \text{ mg L}^{-1}$, being mainly hosted in the sediment precipitating from the Galleria Italia waters. A similar behavior was observed for As. This provides evidence that, despite being in an abandoned Hg-mining district, Hg does seem to significantly impact the Fosso della Chiusa waters, with this creek being fed by the Galleria Italia waters. A detailed investigation on both the dissolved and suspended mercury and sediments along the Fosso della Chiusa creek is presently taking place to better constrain the amount of mercury discharged in the lower reaches of the riverine network. Presently, Galleria Italia is not accessible since it was cemented in the 1980s when the mining activity terminated. However, in order to avoid further flooding events that may have stronger and more destructive impacts that those that occurred in 2009 and 2010, a better water and sediment

management is required. This can only be achieved by restoring access to Galleria Italia. By doing so, the geochemical processes could further be constrained.

Supplementary Materials: The following are available online at https://www.mdpi.com/2076-3 298/8/2/15/s1, Table S1: Description of QA/QC, the criterion of acceptability and the frequency adopted during the analysis of this work.

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